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ANTI-CANCER ACTIVITY OF GREEN THE HESISED SILVER AND GOLD NANOPARTICLES USING CHNA . TUSATA LEAF.

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gree method for the synthesis of silver In this study, we e employe and gold metal noparticles through apid single step method by using Ochna obtusata l extract as a reducing agent. Synthesized AgNP (Silver NP (Go Nanoparticles) were confirmed with SPR Nanopar and eak by UV-Vis spectroscopy. The resulting (Surfa Plas Res nanopticles characterized by employing DLS (Dynamic light ourier Transmission Infra-Red), XRD (X-Ray ΓIR tering), ffraction) **HATEM** (High-resolution transmission electron and (Energy Dispersive X-ray spectroscopy), SAED croscopy). ted Area Electron Diffraction), and Elemental mapping analysis. g, anticancer capabilities of AgNP and AuNP were evaluated Resi against ast cancer cell lines. The results showed that biosynthesized AgNR and AuNP exhibited an absorption peak at 440nm and 540nm, ively. Both SAED and XRD analysis has established the crystalline nature of silver and gold nanoparticles in face-centered cubic (FCC) structure. FTIR spectra haveshown the involvement of plant compounds during the reduction of Ag⁺ and Au⁺ nanoparticles. The TEM images have displayed the spherical structure of AgNP, and an anisotropic structure for AuNP with an average diameter determined to be 26nm and 29nm respectively. Besides, both AgNP and AuNP have exhibited anticancer property against MDA-MB 231 and MCF-7 breast cancer cell lines. In conclusion, we have used O.obtusata for the first time as reducing as well as a capping agent for the synthesis of AgNPs and AuNPs and anticancer activity of AgNPs and AuNPs is established against breast cancer cells.

Keywords: Biosynthesis, *Ochna obtusata*, Metal nanoparticles, Green chemistry, Anticancer.

Introduction

In recent years, Nanoparticles has emerged as a promising tool in drug delivery and cancer therapy^{1.4}. Nanoparticles with 1-100nm size have received massive attention and played a crucial role innanotechnology and nano-medicine due to its salient feature like the high surface area to volume ratio and making them easy to interact with other particles⁵⁻⁷. Nanoparticles disclose unique properties like optical, catalytic, biological, magnetic etc^{8-11} . and therapeutic applications Nanoparticles were synthesized by employing different methods such as synthesis chemical, physical, and biological methods¹²⁻¹⁴. In which, the biological method has a promising effect due to rapid, eco-friendly, and low toxic properties¹⁵. Sources like bacteria, fungi,

algae, plants act as reducing agents in methods¹⁶⁻¹⁸. biogenic Silver nanoparticles synthesize through green method has potent inhibitory and anti-bacterial properties among other nanomaterial¹⁹⁻²¹. Numerous studies on both silver and gold nanoparticles have unique biological revealed their properties²²⁻²⁴ along with anti-proliferative and anti-apoptotic property in can therapy^{3,25,26}. Gold nanoparticles have als received a tremendous interest due to the vast applications in vaccine pro ascarriers, adjuvants, reducing kicity. increasing immunogenic activity, a even in imaging, diagnostic etc^{27-29} . The fore, Green chemistry in nan synth has been increased sign lcantly other biological methode due to it's toxic, cost-effecture, and on-.ph co-fri process³⁰. lical er, toch p n leaves, re , bark, cons ent preser stem, 1 ers act as ducing as well as ts³¹. Flave ids, a class of majorly is olved in the cappin, a pigments an process of grain method³². d is a tree with a woody stem and

D.obtw leaves with attractive yellow ers belows to Ochnaceae family. 0.0 native of East India and commonly known as *Ramdhan champa*³³. are 85 species of Ochna genus widery used as a traditional medicine in Asia, Africa, and America, among which 11 species occur in India³⁴. Plant parts like bark, leaves, and roots of O.obtusata were used in traditional medicine for treating several ailments like dysentery, asthma, diarrhoea, menstrual disorder, cholera, bronchitis. inflammation, and dysmenorrhea^{35,36}. Based on the reported literature, the glossy leaves of *O.obtusata* contains bioactive compounds like flavonoids and bioflavonoids; quercitin 3o-glucoside,kaemperol $3-0-\beta$ -glucoside, ochnaflavonone,2,3-dihydroochnaflavone 7-o-methyl ether,2,3 dihydroochnaflavone³⁷.

In the present study, we approached green chemistry for

th AgNps and AuNps using synthesizin O.obtusata lea aqueous extract (OLAE) as a reducing and abilizing agent. So far to our kno ere is no reported edge, synthesis of articles using nan O.obtusata bio-reductant. as AgN na AuNPis Bi ised hergy dispersive racter by ay S troscopy (EDX), Transmission microscopy (TEM), Selected Area eľ Electric Difference (SAED), Fourier transformer development (SAED), Fourier transformer (FTIR), ynamic Light Scattering (DLS), X-Ray Diffraction (XRD), and Elemental mapping analysis. Further, the anti-cancer property of synthesized nanoparticles were evaluated against breast cancer MDA-MB 231 and MCF-7 cell lines.

Material and methods

O.obtusata leaves were obtained from Tirupati, Andhra Pradesh. An analytical grade Silver nitrate (AgNO₃), Gold chloride trihydrate (HAuCl₄.H₂O) and MTT (3-(4, 5-Dimethylthiazol-2-yl)-2, 5-Diphenyltetrazolium Bromide) was purchased from Sigma-Aldrich (St. Louis, MO). Human adenocarcinoma breast cancer MDA-MB 231 and MCF-7 cell line were purchased from NCCS,Pune. All the aqueous solutions are prepared by using MIilli-Q water.

Preparation of *O.obtusata* aqueous leaf extract:

Thoroughly washed and dried leaves of *O.obtusata* were coarsely ground and stored in an air tight container. The coarse powder of 1g was soaked overnight in 100ml of Milli-Q water. The solution was heated at 40° C under continuous stirring process for 5 min. The solution was filtered by using Whattman No-1 filter paper. Then, the Obtained filtrate was stored a week at 4° C for further experiments.

Synthesis and purification of AgNPand AuNp:

1mM concentration of AgNO3 and HAuCl₄.H₂O solution was prepared. For the synthesis of nanoparticles, we have used different volumes of *O.obtusata* aqueous leaf extract (OAE) (0.5ml, 1ml, 2ml, 3ml, 4ml, 5ml) against constant metal ion concentration. An optimised ratio for the synthesis of AgNP was done by mixing 1ml of OAE with 9ml of AgNO3, whereas AuNP were synthesised by mixing 3ml of OAE with 7ml of 1mM HAuCl₄.H₂O. The reaction mixture for both silver and gold nanoparticles were allowed to stir on a magnetic stirrer at 200rpm speedfor 30 mins incubation at root and temperaturerespectively. Reduction of Ag ions to Ag^0 and Au^+ to Au^0 can be by the change of colour from pal ellow dark brown colour and to urple colourrespectively. Further confirma n of AgNP and AuNP synth don measuring absorption s ctra at nge di 300-700nm at regular interval I.Vsin visible spec Folld scopy. incubation, the tion mixt of ġŇΡ NP ere su cted to cen. and A gation at 1200 n for 30 each by repeated r. The washed washes w Milli-Q 🕅 dried and preserved for pellet wasah aracte tin

Character sation of AgNps and AuNps:

hique optical property of silver he particles was measured by the ion actrum at a range 200-700nm abs using U.v-visible spectroscopy (UV-VISspectrophotometer; Varian Model: 50007. An average hydrodynamic size and Poly Dispersive Index (PDI) of AgNP and AuNP were determined by using particle analyser zetasizer (MALVERN size instruments Nano ZS) at 24.9°C.Bioreduction of silver nitrate by the functional groups present in the leaf extract can be identified by using FTIR measurements (Thermo Nicolet Model:6700 equipped with KBr optics) with the wave number range 5000-700 cm⁻¹ at resolution 0.1cm⁻ ¹.Surface morphology, size, elemental composition and elemental distribution of AgNP and AuNP were determined by using HRTEM, EDX and elemental mapping analysis operated with an instrument (TEM TECNAI-G2 F30-F TWIN) at 200kV.Preparation of sample for

was done by dropping 5µl TEM analy of ethanol disp ed AgNP and AuNP on a carbon coated econor grid and allowed to dry under The crystalline ld nanoparticles structure of ilver and was analyse by the D and XRD by using XRD record RD pa b) with the step N ana ical operated with CuK α , at settings 0.0230 40kV, der 2 Θ range from 20° size of oth AgNP and AuNP was 80° e Debye-Scherrer equation. calculate *a-vitro* anti-cancer activity of AgNP and

Cen culture conditions:

Cancer cell lines MDA MB-231, MCF-7 (human breast adenocarcinoma epithelial cells) were cultured in Dulbecco's Modified Eagle Medium (DMEM) with 1000mg/L glucose, Lglutamine and sodium bicarbonate which is supplemented with 10% FBS, 1% Penicillin (100U/ml) and streptomycin (100ug/ml). Cultured cells were incubated under 5% Co₂ at 37°C until it reaches 80% confluence.

MTT assay:

Cytotoxicity evaluation of synthesised AgNP and AuNP was performed by using MTT (3-(4, 5-Dimethylthiazol-2-yl)-2, 5 Diphenyltetrazolium Bromide) assay as previously reported protocol³⁸. All the experiments were done in triplicates. Cell viability and 50 % growth inhibition (IC₅₀) were determined and calculated by the following equation.

% of cell inhibition =100-[(At-Ab)/ (Ac-Ab)] x100 and

% of cell viability = [(At-Ab)/ (Ac-Ab)] x100 DNA Fragmentation assay:

DNA fragmentation is largely considered as a distinctive feature of apoptosis³⁹. DNA fragmentation assay is performed as described byHackenberg, S. et al. 2011^{40} , with slight modifications. MCF7 (1*10⁶ cells/mL) were seeded in 6-well microplates and treated with increasing concentration 5, 10 and 15μ g/mL of AgNP and AuNP respectively

for 24h. Following cells were collected and DNA is isolated by using DNeasy blood and tissue kit (Qiagen). The resultant DNA was quantified using nanodrop. For Estimating the DNA fragmentation equal amounts of DNA is loaded and run on 1.0% agarose gel containing 1 μ g/mL ethidium bromide at 90 V for 1.5 h in 1 × TAE buffer of pH 8.5. Laterthe DNA fragments v visualized by exposing the gel ultraviolet light, followed by photography. **Results and discussion**

Synthesis of AgNP and AuNPs:

In this study for the first time, e are reporting biosynthesis of AgNP and AuNPs by utilising aq f ex O.obtusata as a reduci agent rapic single pot green rnthesis ar An 6à overview work an ofsilv and d nanoparticles esisis re sent in (fig: silver gold)).Bo es synthe was observed and nanopa confirmed the colour range of reaction mixture to a brown and purple colour ely du to Surface Plasmon (SPR)⁴ Resona

Investigation of results has aled that O.obtusata aqueous leaf extr as stabinging agent for the synthesis of AgNps and AuNps Synthesis of Aging and AuNP was confirmed within 6 hrs and 30 min respectively, with a colour change from pale yellow to dark brown and purple respectively. Reduction of Ag⁺ ions to Ag⁰ (AgNps) and Au⁺ to AuNps were confirmed by using U.V visible spectroscopy. Surface plasmon resonance (SPR) absorption band^{42,43} is observed at 440nm for AgNPs and 540nm for AuNps(fig: 1,(II) (a) and (III)(a)). Time dependent synthesise of AgNps and AuNps were recorded using measuring SPR absorption band as shown in UV-VIS spectra in fig:1 (fig:1,(II) (b) and (III) (b))The intensity and broadening of SPR bandwas increased constantly by the increase of reaction time due to the

increase h the concentration of nanoparticles

Characterisation gNP and AuNps: DLS:

verage hy dynamic size of The JP measure was 86.79nm rectivel which is based AgNP and A an nm \ h DLS particle scatte 10 he PDI (Polydispersity Index) lyser AgNP and AuNP was 0.346% and vž 0.29 respe vely which represents preparation and acceptable monod drug delivery applications⁴⁵. The size distribution intensity of synthesised Gold and Silver was shown in for fig: 1 (ii); (a) and (b).

TEM, EDX and Elemental mapping analysis:

Synthesised AgNP and AuNP were observed under TEM for visualizing their size and morphology. Results interestingly revealed that gold nanoparticles show shape like anisotropic hexagonal, triangular, rod and spherical particles with an average diameter ranging 26nm (figure: 2 (I);(a). Whereas, silver nanoparticles have shown spherical shape with an average diameter ranging 28nm (figure: 2 (I); (b)). Besides AgNP and AuNP have shown uniform and monodispersed in nature respectively.

The crystallinity and the elemental distribution inside the AgNP and AuNP nanoparticles were analysed in more detail by energy dispersive X-ray spectroscopy. A characteristic sharp absorption peaksfor AgNP(fig:2 (II);(a)) and AuNP(fig:2 (II);(b)) at3keV and 2.5keVwas observed respectively in EDX spectrum as shown in fig:2 (II);(a & b), which is a typical energy value for metallic silver and gold nano crystallites⁴⁶. Additional peaks of carbon and oxygen were also present confirming the successful capping of *O.obtusata* plant compounds on the surface of AgNP and AuNPs. Furthermore, absorption peak for copper is also observed due to the usage of carbon-coated copper grid for the study⁴⁷. Further, an elemental mapping analysis of AgNP and AuNP were carried out to know



Figure 1:

(I) Schematic synthesis of Silver and Gold nanoparticles using *O. obtusata* Aqueous leaf extract (II) (a) U.V-Vis spectra of AgNO3 Solution, *O. obtusata* aqueous leaf extract (OAE) and silver nanoparticles (b) Time dependant analysis of silver nanoparticles (AgNP'S) prepared by 9ml of 1mM silver nitrate solution and 1ml of plant extract

(III) (a) U.V-Vis spectra of HAuCl4.H2o Solution, *O.obtusata* aqueous leaf extract (OAE) and gold nanoparticles (b) Time dependant analysis of Gold nanoparticles (AgNP'S) prepared by 7ml of 1mM silver nitrate solution and 3ml of plant extract.

(IV)Size distribution by intensity of (a) Gold nanoparticles and (b) Silver nanoparticles.



Figure 2:

(I) High-Resolution Transmission Electron Microscopy (HRTEM) images of biosynthesised gold (a) and silver (b) nanoparticles at 100nm, 50nm and 10 nm resolution.

(II) EDX spectra of silver (a) and gold (b) nanoparticles.

(III) Selected area elemental mapping results indicate the distribution of elements, the TEM micrograph of silver nanoparticle pellet solution (a, b), and silver element; blue (c) respectively. TEM micrograph of gold nanoparticle pellet solution (d, e), and gold element; blue (f) respectively.

FTIR, XRD spectra and SAED pattern:

FTIR spectra of *O. obtusata*, AgNP, and AuNP was shown in fig:3 (I) revealing the functional groups involved in reduction and stabilisation. The IR spectra of *O.obtusata* showed a wide range absorption peaks around 3407,2919, 2851, 1521,1284,1064,589 cm⁻¹ assigned to O-H, C-O, C-H, C=C, S=O and C-Br which are vibrational-stretching of alcohols, phenols, tertiary amines, alkanes, and alkenes respectively⁴⁹. The IR spectra of AgNP has displayed a prominent absorption peak at 3409 cm⁻¹ corresponding to O-H stretching and peaks at 1611,1526,1441,1270 cm⁻¹ indicates the involvement of flavonoids, phenols, alkanes, and alkenes in the reduction of

(111),(200) (220) facets of silver nitrate. Whereas IR spectra of AuNP corresponds showed an intense peak at 3397 cm⁻¹ ic (FCC) crystal lattice face centred structure which matched with earlier to O-H stretching assigned and sis method⁴⁷. The 1610,1513,1441,1283,514 cm⁻¹ assigned to reported g SYL C=C, N-O,C-H.C-O,C-Br vibrational calculated oic struc of AgNP and stretching respectively indicating the AuNP are 0.5nm ai 10.2nm size using following involvement of alkanes, alkenes, amines, ned by re alcohols, and halo compounds during oye-So reduction. Herewith the IR spectra show e size.D = that O. obtusata leaf extract acts reducing as well as capping agent durin her indicates the mean silver and gold nanoparticle synthesis. e,K is a scherrers constant crystallin Powder XRD pattern of AgNP and $(=0.94),\lambda$ is X-ray wavelength, β is full was shown in fig:3 (ii); (a,b). tained If maximal (FWHM) intensity, θ is diffraction peaks of AgNP and Au at 20 Bragg angle. degrees values of 37.92°__44.05° 4.6° **(I**) (b) 110 (111) . 444 1 128 20 (degrees Wavenumber (cm⁻¹) ഷ (a) (b) 5 1/n (d) (c)

Figure: 3 (i) FTIR spectra of *O. obtusata* leaf extract and silver and gold nanoparticles. (ii) ;(a & b) indicates XRD spectra of silver (a) and gold nanoparticles (b). (iii) SAED pattern of silver (a) and gold (b) nanoparticles where the four circular rings assigning to (111), (200), (220), and (311) characterise the face cantered cubic structure which was further proved with a lattice fringe spacing in fig:3 (iii);(c & d) with the distance of 0.26nm and 0.23nm obtained from high resolution transmission electron micrograph (HRTEM). Through the results obtained from XRD, SAED and HRTEM micrograph images, the synthesised silver and gold metal nanoparticles from *O. obtusata* are crystalline in nature.

Anti-cancer activity of AgNP and AuNps:

In-vitro cytotoxicity of AuNP and AgNP against MCF-7 and MDA-MB-231 cells were measured using MTT colorimetric assay after 48h incubation. Results of MTT assay were shown in fig: 4, untreated cells were considered as control. In agreement with other studies⁵⁰⁻ ⁵², our synthesised nanoparticles showed significant cytotoxicity compared to the



ure: 4 Dec dependant in-vitro Cytotoxicity assay of Gold nanoparticles (A, B) and silver nanoparticles

Fragmentation assay:

Apoptosis inducing capabilities of synthesised AgNP and AuNP were verified by DNA laddering assay. The results revealed that AgNP and AuNP have inter nucleosomal induced DNA fragmentation in MCF7 cells with increasing concentration of nanoparticles (fig 5), DNA fragmentation is the characteristic feature of apoptosis induction. When compared to the control cells (untreated) the extent of DNA fragmentation caused by AgNP is prominent with increasing concentration. Control cells have shown minimum breakage of DNA' whereas at 15 µg/mL of AgNP and AuNP has exhibited there by extensive double strand breaks, thereby yielding a ladder appearance (fig 5).

Conclusion:

The leaf extract of O. obtusata acts as a reducing agent as well as stabilising agent for the synthesis of Silver and Gold nanoparticles. An ecofriendly and fast facile synthesis of AgNPs and AuNPs by O. obtusata leaf extract were done. Synthesised AgNP and AuNP are stable for 2 months without the involvement of anv hazardous chemicals further characterisation was done by FTIR, TEM. DLS. SAED, XRD. and elemental mapping analysis. Ouranticancer activity assay revealed that AuNP and AgNP'S had exhibited significantly high toxicity towards breast cancer cell lines MDA-MB-231 and MCF -7 than plant extract alone.



Figure 5: DNA fine entation and sis using agarose gel electrophoresis. Lane 1- Ladder, Lane 2- control (untreated cells), Lane 3- AgNP- 5 will one 4- AgNP- 10ug/ml, Lane 5-AgNP- 15ug/ml, Lane 6-AuNP- 5 ug/ml, and Laus-AuNP- 15 ug/ml.



Graphical abstract: Schematic synthesis of Silver and Gold nanoparticles using O. obtusata aqueous leaf extract.

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